

The chemistry of the massive outflow source G5.89–0.39

I. A 330 – 360 GHz molecular line survey

M.A. Thompson and G.H. Macdonald

Electronic Engineering Laboratory, University of Kent, Canterbury, Kent, CT2 7NT, UK

Received October 15; accepted November 27, 1998

Abstract. We have performed a 330 – 360 GHz molecular line survey of the molecular gas associated with the ultra-compact (UC) HII region G5.89–0.39. 142 lines originating from 19 species were detected and a further 6 lines that could not be identified. Over 50% of the identified molecular lines are from sulphur-bearing species, with SO₂ (and its isotopomer ³⁴SO₂) exhibiting the most lines. No emission from heavy organic molecules (e.g. dimethyl ether, ethanol, methyl formate) was detected. The lines can be divided into two types by their profiles: narrow lines with no (or slight) wing emission which probe a molecular envelope surrounding the UC HII region and broad lines with considerable wings which may probe the gas in the bipolar molecular outflow associated with the UC HII region. We discuss the chemical models which may explain the features observed in the survey and outline the strategy for the chemical modelling of the evolution of the molecular gas, to be considered more quantitatively in a following paper.

Key words: ISM: individual objects: G5.89–0.39 — ISM: molecules — ISM: jets and outflows — line: identification — ISM: abundances — radis lines: ISM

1. Introduction

Outflows play a fundamental role in star formation. Excess angular momentum in the protostellar system is removed by the outflow, the star-forming core of dust and gas is disrupted by the outflow, possibly limiting the mass of the resulting star, and the interaction of the outflow with the natal molecular cloud may have far-reaching effects on subsequent star formation within the cloud.

Outflows from low mass young stellar objects (YSOs) have been well studied to date; for example the reviews by Bachiller (1996) and Fukui (1993) and references therein. Molecular outflows and high-velocity ionised jets have

been observed toward low mass YSOs. The molecular outflows are thought to be driven by the jets, with the momentum from the jet transferred to the molecular material by entrainment (either prompt entrainment by bow shocks at the head of the jet or steady state entrainment by turbulent mixing along the sides of the jet, De Young 1986). There is a strong relationship between the mechanical force and luminosity of the molecular outflow and the bolometric luminosity, particularly the 6cm continuum emission (Cabrit & Bertout 1992) of the central protostar driving the outflow, which suggests a close association between ionised and molecular gas. The driving force behind the bipolar ionised jets is not yet clear. Current theories suggest the jets are wind-driven, possibly by either a disc-collimated wind (Pelletier & Pudritz 1992) or from boundary layer effects in the protostellar accretion disc (Shu et al. 1995).

The overall picture of massive YSO outflows is much less certain. The large distance of most high mass star forming regions makes it difficult to resolve fine detail and the higher degree of turbulence present in massive star forming cores confuses the location of high-velocity emission from outflows. It is important though to determine the properties of massive YSO outflows to place them in the context of the models developed for low mass YSO outflows and to test the predictions of the models against the properties of high mass YSO outflows. Until recently only a few isolated examples of massive stellar outflows were known (e.g. G5.89–0.39, DR 21 and G45.12). Studies by Shepherd & Churchwell (1996a,b) have revealed high velocity CO emission present in 90% of a survey of massive YSOs and have determined the properties of five outflows by mapping a sample of sources with high velocity emission from the survey.

One of the most important aspects of this recent work on massive YSO outflows has been the confirmation that massive YSO outflows lie at the more extreme end of the scale. The mass contained in massive YSO outflows is at least an order of magnitude higher than that of most YSO

outflows. The flow rates of massive YSO outflows are also in general larger than their low mass counterparts. If both outflow phenomena share a common origin, massive YSO outflows thus represent an ideal test of the models developed for low mass YSO outflows. Doubts have been raised whether disc-collimated wind or boundary-layer (so-called X-wind models) models of outflows are capable of driving the large masses in massive YSO outflows (Churchwell 1997), although the massive bipolar outflow associated with G192.16 exhibits properties of both disc-collimated and X-wind models (Shepherd et al. 1998). Further studies of massive YSO outflows are necessary to determine whether high and low mass protostars share a common outflow mechanism.

The chemistry in massive and low mass YSO outflows is presumably driven by a common process. Outflows drive shocks into the ambient molecular gas and strongly affect the chemistry. The chemistry is altered by both the heating action of the shock which permits reactions with high activation energies to take place and by the injection of disrupted grain mantles (and perhaps grain cores) into the gas phase. The most striking difference between quiescent cloud and shock-driven chemistry is in the abundance of SiO, which is several orders of magnitude higher toward shocked regions (Schilke et al. 1997a). Other species, such as SO, NH₃ and CH₃OH are also observed to have raised abundances toward outflows (Bachiller 1996 and references therein).

Molecular line surveys of low mass YSO outflows (e.g. Blake et al. 1995; Bachiller & Perez-Gutierrez 1997) have been extremely useful in characterising the chemistry of the outflow and in deriving the physical parameters of the gas from the detected molecular lines. However as in the dynamical study of outflows the main effort has focussed on low mass YSO outflows. For this reason we have undertaken a 330 – 360 GHz molecular line survey of the massive YSO outflow associated with G5.89–0.39. The outflow is compact, the lobes are separated by $\sim 6''$ as traced in SiO emission, and massive with an outflowing mass of ~ 80 solar masses (Harvey & Forveille 1988; Acord et al. 1997). There is also a dense envelope of dust (Harvey et al. 1994) and molecular gas (Gomez et al. 1991) surrounding the UC HII region, which is the prototypical example of the shell morphology (Wood & Churchwell 1989).

The compactness of the outflow means that both lobes and the envelope gas associated with the UC HII region can be sampled in a single $13''$ HPBW observation. The line profiles can be used to determine whether emission arises predominantly from the envelope gas located close to the UC HII region or from within the accelerated gas of the outflow. Turbulence within the envelope, observed to be fairly ubiquitous towards regions of massive star formation (Plume et al. 1997), will confuse the issue although the line profiles of particular species will indicate species that can be explored further with higher resolution interferometric observations.

The HII region G5.89–0.39 is also postulated to be young from observations of its expansion rate (Acord et al. 1998) and the high dust density in the immediate environment (Harvey et al. 1994). We will be able to contrast the results of the 330 – 360 GHz survey with other recent molecular line surveys of G34.26 (Macdonald et al. 1996) and Orion-KL (Schilke et al. 1997b), in order to test the hypothesis that the chemistry of these objects can be used as a “chemical clock” to date the evolution of the molecular gas.

In summary, the aims of this survey are: (1) to characterise the chemistry of the molecular gas associated with G5.89–0.39, (2) to identify species located within the outflow by their high velocity line wings, (3) to derive the physical parameters of the molecular gas and (4) to examine whether hot core or shock-driven chemical models can reproduce the observed column densities. Parts (1) and (2) are presented in this paper, along with a first approach to part (3) by deriving the rotational temperatures and column densities of those species with sufficient detected transitions. A more detailed physical analysis and chemical modelling (parts 3 and 4) will be contained in the forthcoming Paper II.

The observations and data reduction procedure are detailed in the next section. The identified lines, inspection of their profiles for high velocity emission and rotation diagrams are presented in Sect. 3. The implications of the survey are discussed in Sect. 4 with particular reference to shock-driven and hot core chemical models.

2. Observations and data reduction

2.1. Observations

The observations were made with the James Clerk Maxwell Telescope (JCMT)¹ between the 8th and 14th March 1996. All observations were made at the coordinates $\alpha(1950) = 17^{\text{h}} 57^{\text{m}} 27^{\text{s}}$ and $\delta(1950) = -24^{\circ} 03' 57''$, which is the estimated centre of the UC HII region (Wood & Churchwell 1989). The pointing accuracy of the telescope was checked regularly against the peak continuum position of the UC HII region G34.26+0.15 and was found to be good to within $5''$. It was found that beam-switching (i.e. chopping the secondary mirror from on-source to off-source) was much superior to position switching for obtaining extremely flat baselines. A chop throw of $3'$ in RA was used to keep a constant reference position, with a chopping frequency of 1 Hz. $3'$ was more than sufficient to avoid contamination in the reference position for all species except CO (as can be seen in Fig. 1).

¹ The JCMT is operated by the Royal Observatories on behalf of the Particle Physics and Astronomy Research Council of the UK, the Netherlands Organisation for Scientific Research and the National Research Council of Canada.

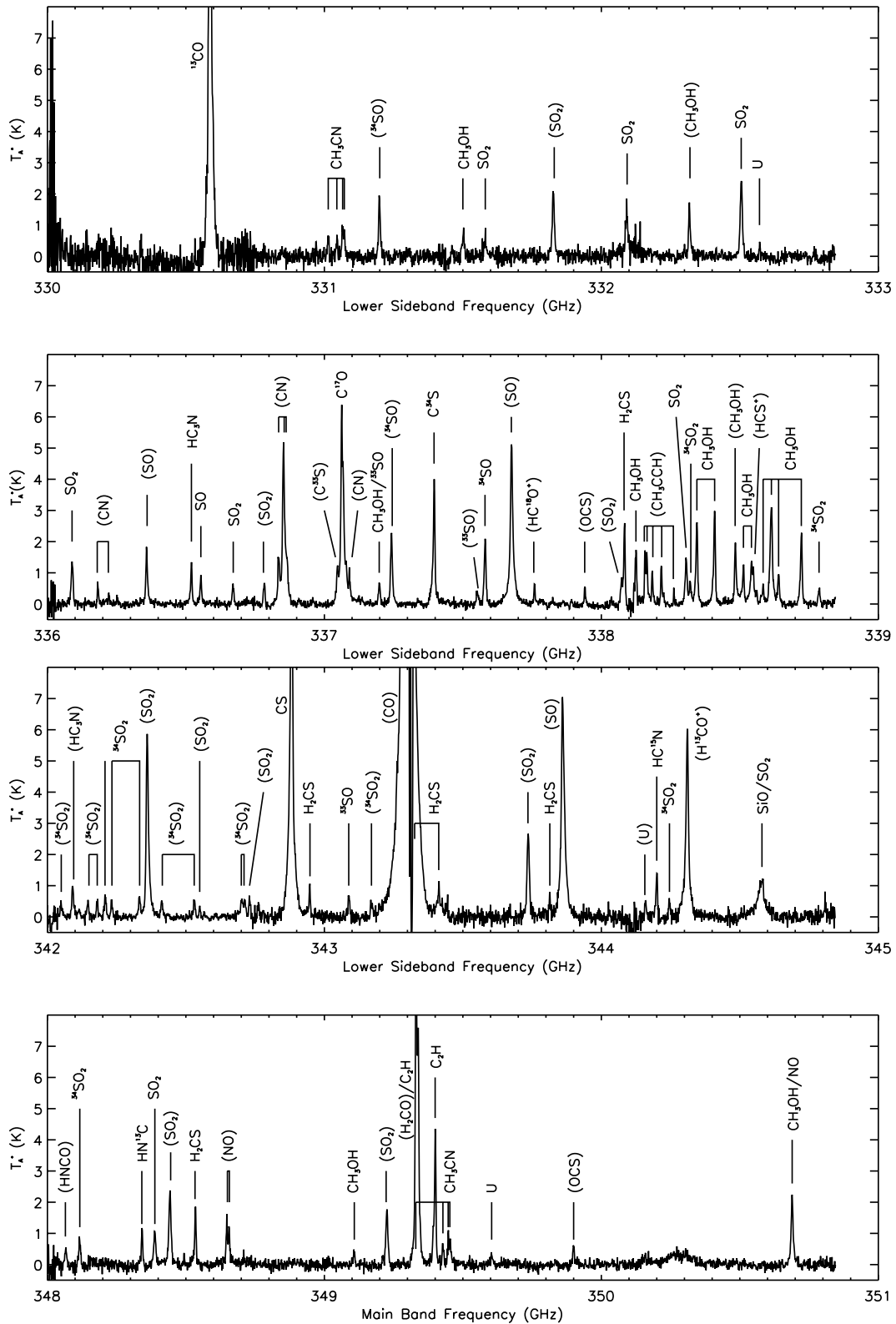


Fig. 1. Spectra observed during the 330 – 360 GHz survey of G5.89–0.39. The frequency scale given for each block of spectra is the lower sideband scale, the upper sideband scale is omitted as it is not continuous for the blocks of concatenated spectra. Species identifications are shown for each line and upper sideband lines are indicated by brackets

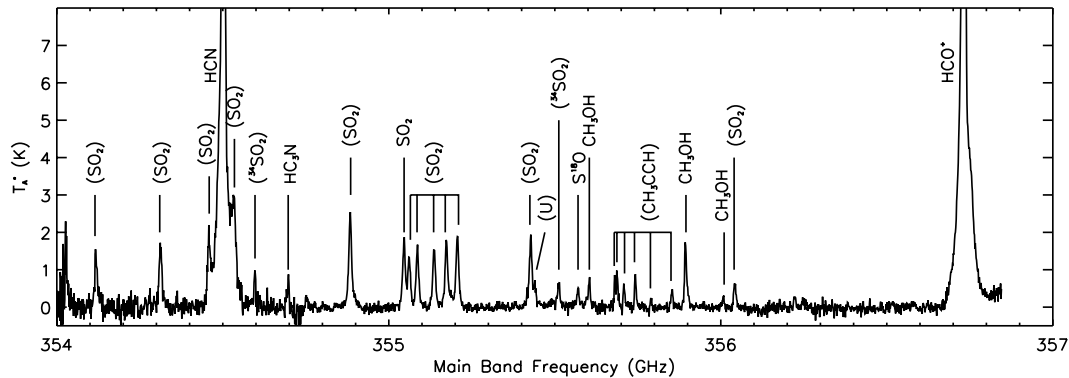


Fig. 1. continued

To cover the frequency range of the survey the 345 GHz SIS junction receiver B3i (RxB3i) was used in conjunction with the Dutch Autocorrelation Spectrometer (DAS). The DAS was used in 760 MHz bandwidth mode and with RxB3i as a frontend produces dual sideband spectra. Dual sideband spectra comprise two frequency bands (the upper and lower sidebands) folded over one another to produce a composite spectra. The upper and lower sidebands are separated in frequency by approximately twice the local oscillator intermediate frequency (IF), depending on the doppler correction for the source velocity. The upper sideband frequency scale is reversed relative to the lower sideband scale. The velocity of G5.89–0.39 with respect to the Local Standard of Rest (V_{LSR}) was assumed to be $+9.4 \text{ km s}^{-1}$. For RxB3i the IF is 1.5 GHz and the upper and lower sidebands are separated by approximately 3 GHz. Each spectrum taken thus represents a total frequency range of ~ 1.5 GHz and this was used to reduce the total number of spectra needed to cover the frequency range of the survey.

The spectra were all observed with the “main band” set to the lower sideband, which means that the other (upper) sideband covers a frequency range of the same width roughly 3 GHz higher in frequency. We took spectra with their central frequency incremented by 700 MHz (ensuring an overlap of 30 MHz between spectra) until the lower sideband had covered the first 2.8 GHz of the frequency range. The upper sidebands of these spectra cover the next 2.8 GHz of the frequency range with a 200 MHz gap in coverage. This block of 4 spectra thus covers a total frequency range of 5.6 GHz. The remaining parts of the frequency range were observed in the same manner. The 200 MHz gaps between the blocks of spectra were to be covered by additional spectra taken at the end of the observing run, however due to bad weather this was not achieved. These gaps do not contain many lines of significance; searches of spectral line catalogues (both of predicted and observed lines) indicate that few lines occur in these frequency ranges. The blocks of spectra (with individual spectra concatenated) are shown in Fig. 1.

Two problems inherent in dual sideband spectra are the allocation of features to a particular sideband (i.e. upper or lower) and the possible overlapping (blending) of lines from each sideband. To determine the sidebands (and hence line frequencies) extra spectra with a local oscillator shift of $+10$ MHz were taken. In the shifted spectra lines in the upper sideband will appear to shift frequency by 20 MHz relative to those in the lower sideband. Blended lines from both sidebands were separated by this technique whenever possible.

With the DAS in 760 MHz mode the spectral resolution is 0.756 MHz. Each spectrum was divided into channels of 0.625 MHz, although later in the data reduction process all spectra were binned to a channel width of 1.25 MHz to improve signal to noise. The standard chopper-wheel calibration method of Kutner & Ulich (1981) was used to obtain line temperatures on the T_{A}^* scale, i.e. corrected for the atmosphere, resistive telescope losses and rearward spillover and scattering. T_{A}^* can also be corrected for forward spillover and scattering to give the corrected receiver temperature T_{R}^* where $T_{\text{R}}^* = T_{\text{A}}^* / \eta_{\text{fss}}$ and η_{fss} is the forward spillover and scattering efficiency (0.7 for RxB3i at 345 GHz).

2.2. Data reduction and line identification

The data were reduced using the Starlink spectral line package SPECX. Linear baselines were subtracted from the spectra and the line parameters of peak temperature (T_{A}^*), central frequency ($\nu(\text{obs})$) and line width at half maximum ($\Delta\nu_{1/2}$) were measured. Values for the noise in the spectra were evaluated using line-free channels and the typical rms noise level was found to be $\sigma \sim 0.1$ K at a spectral resolution of 1.25 MHz. Features below a detection limit of 5σ were ignored to avoid inaccurate line identifications. These data are given in Table 5 for each line. Multiple independent detections of each line (i.e. including the 10 MHz shifted spectra) are listed.

The identification of molecular lines detected in this survey was achieved by comparing their central frequencies primarily with the JPL spectral line database

Table 1. Species (including isotopomers) identified in the survey. The identification of certain species from single line detections must however be regarded with caution

Species	Number of detected lines	Species	Number of detected lines
CO	1	HCN	1
¹³ CO	1	HC ¹⁵ N	1
C ¹⁷ O	1	HN ¹³ C	1
SO	5	HNCO	1
³⁴ SO	3	NO	2
³³ SO	3	HCO ⁺	1
S ¹⁸ O	1	HC ¹⁸ O ⁺	1
SO ₂	32	HC ₃ N	3
³⁴ SO ₂	19	CH ₃ CCH	11
C ₂ H	2	CH ₃ CN	8
CS	1	CH ₃ OH	18
C ³⁴ S	1	H ₂ CO	1
C ³³ S	1	H ₂ CS	6
OCS	2	U	6
CN	8		

(Poynter & Pickett 1985). Other lists used include Lovas (1992), the methanol lists of Anderson et al. (1993) and the observational lists of lines seen by Jewell et al. (1989) toward Orion-A, Schilke et al. (1997b) toward Orion-KL and Macdonald et al. (1996) toward G34.26+0.15.

3. Results

3.1. Identified lines and their profiles

142 lines originating from a total of 19 species were detected and are shown in Table 5. 6 further lines could not be identified and are indicated as such by *U* in Table 5. Many isotopomers were detected, particularly the ³⁴SO₂ isotopomer. A list of the identified species (including isotopomers) and the total number of lines from each is given in Table 1. Approximately 50% of the identified lines originate from sulphur-bearing species. Many of these lines (e.g. CS, SO, SO₂ and to some extent C³⁴S and ³⁴SO) exhibit broad line wings suggesting that their emission originates from regions of wide velocity dispersion, perhaps from the molecular outflow. Other non sulphur-bearing species known as typical outflow tracers (CO, HCO⁺, CH₃OH) also display a wide range of velocities. Isotopomers such as ¹³CO, H¹³CO⁺ and C³⁴S have broad line wings, which indicates that the outflowing gas has a significant optical depth in these molecules. The density tracer HCN also appears to have broad line wings, although this is complicated by the presence of two image band SO₂ lines blended with the HCN 4–3 line. SiO exhibits extremely broad emission (across ~ 50 km s⁻¹), however it is again unfortunately blended with a vibrational SO₂ line in the image band.

In contrast to the broad lines thought to have an origin within the outflow, several species with narrow line profiles (e.g. HC₃N, CH₃CN, CH₃CCH, NO) trace a relatively quiescent region of gas, probably the molecular envelope surrounding the UC HII region (Gomez et al. 1991).

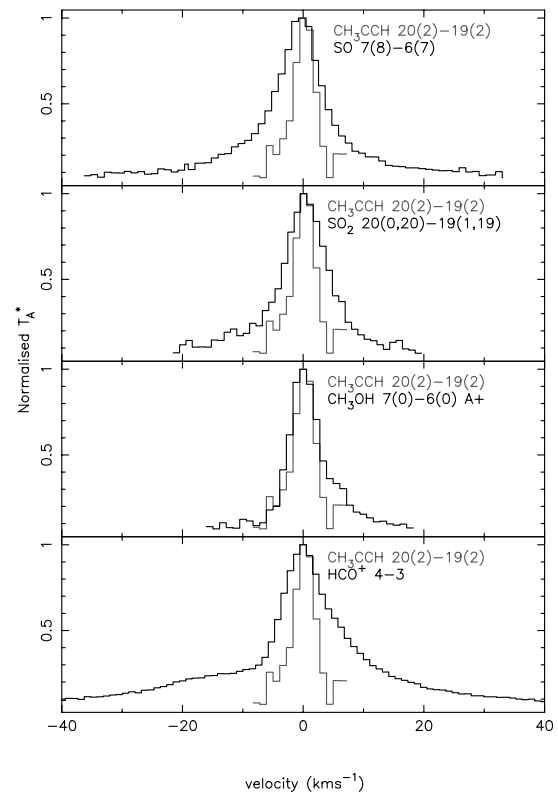


Fig. 2. Spectra of broad lines from species known to probe outflows are compared to the CH₃CCH 20(2) – 19(2) line. Broad red and blue-shifted wings are clearly visible in the SO, SO₂ and HCO⁺ lines, whereas the CH₃OH 7(0) – 6(0) A+ line only clearly exhibits a red-shifted wing

Figure 2 shows a selection of high-velocity lines compared with the CH₃CCH 20(2) – 19(2) line as representative of the low velocity dispersion lines. The temperature scale is normalised to highlight the differences between them. The differences between the narrow CH₃CCH line which probes the gas of the molecular envelope and the broad lines of the outflow tracers SO, SO₂ and HCO⁺ are obvious. The CH₃OH line also has a distinct red wing.

In contrast to other high-mass YSO sources such as G34.26+0.15 and Orion-KL, there is no observational evidence of the presence of heavy organic molecules such as ethyl cyanide (CH₃CH₂CN), methyl formate (HCOOCH₃) or dimethyl ether (CH₃OCH₃). These molecules have large numbers of emission lines detected towards the other sources and are evidence of a complex hot core chemistry, which may be absent in the molecular gas associated with G5.89–0.39.

3.2. Integrated intensities

We have also compared the total integrated intensities from each molecule, in a similar manner to that of the Orion-KL survey (Schilke et al. 1997b). It is interesting to compare the results from G5.89 with those of Orion-KL. In Orion-KL the most dominant molecule in terms of

Table 2. Integrated intensity for each species. The integrated intensities have been evaluated using the T_A^* scale. Where it has not been possible to place a firm value on the total integrated intensity due to blended lines an upper limit is indicated

Molecule	Total integrated intensity (K MHz)
CO	<2049.8
SO ₂	<829.2
SO	514.9
HCO ⁺	462.5
HCN	<340.6
CH ₃ OH	<300.5
CS	251.6
CN	189.7
H ₂ CO	120.6
C ₂ H	93.9
CH ₃ CCH	81.0
H ₂ CS	64.8
SiO	<40.0
CH ₃ CN	35.4
HC ₃ N	29.1
NO	23.2
HNC	13.0
OCS	10.8
HCS ⁺	7.9
HNCO	7.2

integrated intensity is SO₂ which has a total integrated intensity more than twice that of CO. In G5.89–0.39 the most dominant molecule is CO with twice the integrated intensity of SO₂. This is due to beam dilution, since G5.89–0.39 is approximately 4 times more distant than Orion KL. The same effect can be seen in other species such as CH₃CCH and CH₃CN; in Orion-KL the former is much less important for cooling the gas than the latter. However in G5.89 CH₃CCH has twice the integrated intensity of CH₃CN. This is probably due to the property of CH₃CCH of tracing more extended gas (by virtue of its lower dipole moment), making CH₃CCH less affected by beam dilution. Care must be taken in interpretations such as this though, as genuine chemical differences may affect the relative abundances of particular species between the two clouds and hence their integrated intensities.

3.3. Rotation diagrams and lower limits to column density

The rotation diagram approach assumes optically thin gas in LTE, for which the (beam-averaged) column density (N) can be written as:

$$N = \frac{3k}{8\pi^3} \frac{\int T_R dv}{\nu S \mu^2 g_l g_K} Q(T_{\text{rot}}) \exp\left(\frac{E_u}{kT_{\text{rot}}}\right) \quad (1)$$

where $\int T_R dv$ is the integrated intensity of the line, ν is the line frequency, S is the line strength, μ is the permanent electric dipole moment, g_l and g_K are the reduced nuclear spin degeneracy and the K-level degeneracy of the molecule respectively. E_u is the energy of the

upper level of the line and T_{rot} is the rotational temperature of the molecules comprising the gas and $Q(T_{\text{rot}})$ is the corresponding partition function. Values for $Q(T_{\text{rot}})$ were obtained by interpolating the values given in the JPL molecular line database to the appropriate temperature. Equation (1) was rearranged to an equation for a straight line and T_{rot} and N for each species were determined by a least-squares fit. A rotation diagram for ³³SO was not constructed owing to the lack of well-determined molecular parameters, such as the partition function. Lower limits to the column densities (N_{min}) of the remaining species were evaluated using the minimum point of Eq. (1), which occurs at $T_{\text{rot}} = E_u/k$ for linear molecules and $T_{\text{rot}} = \frac{2}{3}E_u/k$ for symmetric and asymmetric top molecules. Both methods are described more fully in Hatchell et al. (1998a).

The results of the rotation diagram and lower limit analyses are given in Tables 3 and 4. Abundances have been calculated for the rotation diagram species by assuming a spherical cloud with a radius of 0.2 pc and a number density $n(\text{H}_2) = 10^4 \text{ cm}^{-3}$ (taken from Gomez et al. 1991). The abundance of ³⁴SO₂ indicates that the SO₂ abundance may be underestimated due to optical depth effects and should be closer to 3×10^6 . The rotation diagrams are shown in Fig. 3. Certain lines have been excluded from the analysis; self-absorbed and blended lines, and those lines that could not be unambiguously identified with a single species. The data from SO, ³⁴SO and H₂CS did not give a satisfactory fit to a straight line and lower limits to column density have been evaluated for these molecules.

The rotation diagrams indicate that the molecular gas associated with G5.89 has a temperature of roughly 60–70 K and a column density of $\sim 10^{15} \text{ cm}^{-2}$. SO₂ appears to probe somewhat hotter gas with temperature in excess of 100 K. The temperature for the less optically thick ³⁴SO₂ is higher than that of SO₂ which is consistent with (although not conclusive evidence for) the temperature of the molecular gas increasing towards the centre.

4. Discussion

The results of the survey show that the molecular gas associated with G5.89 does not have such a complex chemistry as other massive YSOs (G34.26+0.15 and Orion-KL). Roughly an eighth of the number of lines seen toward Orion-KL (Schilke et al. 1997b) and a half of the number of lines detected toward G34.26+0.15 (Macdonald et al. 1996) were seen in this survey. 19 species were detected, mostly sulphur-bearing, as opposed to the 35 species seen in the G34.26 and Orion-KL surveys. No lines of heavy organic species (e.g. CH₃OCH₃, HCOOCH₃, CH₃CH₂CN) were detected. We note that $\lambda = 1.3 \text{ mm}$ lines of HCOOCH₃ and CH₂CHCN were detected toward G5.89 by Acord et al. (1997), the prevalent lines of these species seen in the 330–360 GHz range toward Orion-KL and G34.26+0.15 were not detected in this survey. This may be due to the sensitivity limit of this survey, the lines of these species detected in G34.26 are often weak with

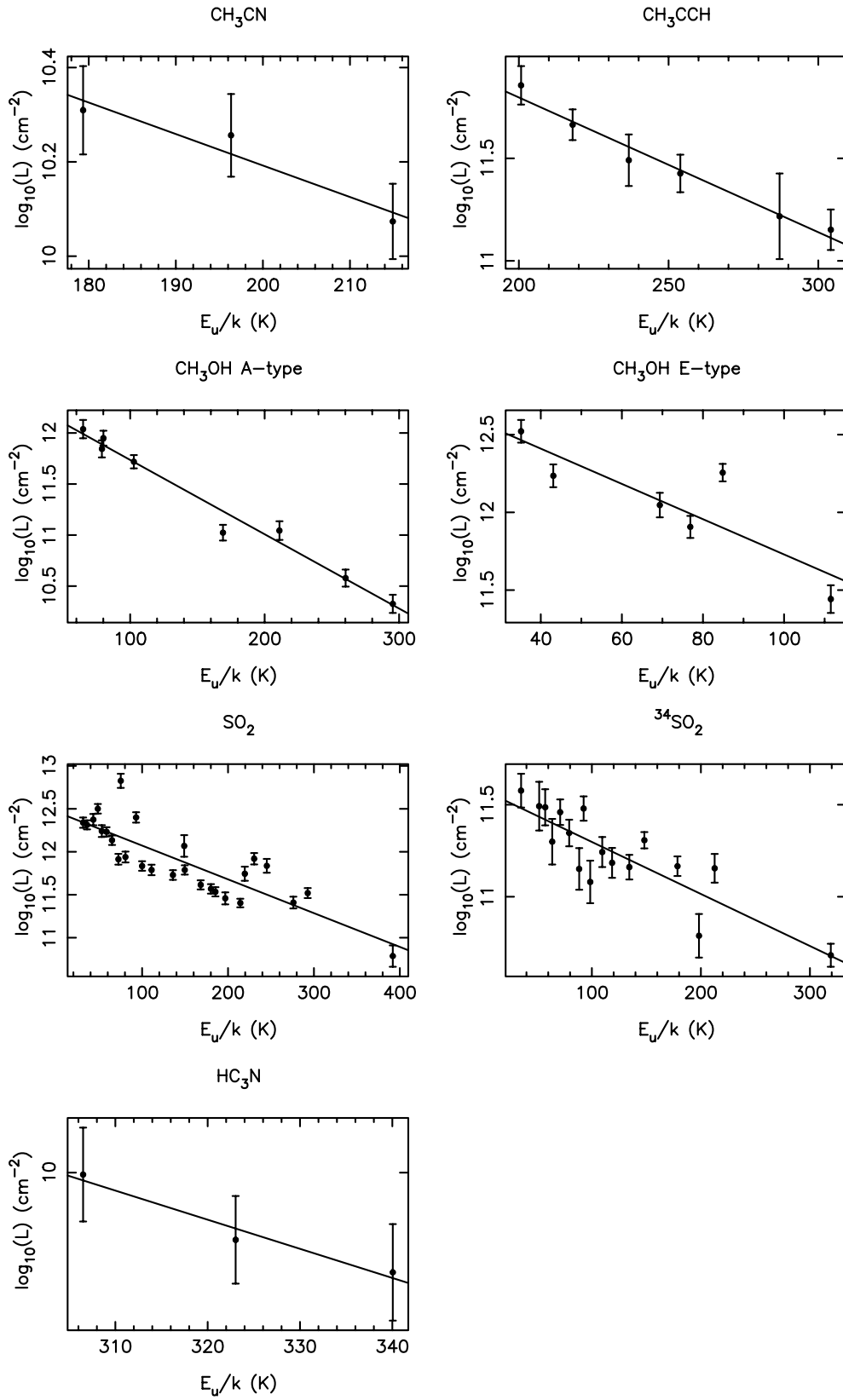


Fig. 3. Rotation diagrams for species with sufficient detected lines

Table 3. Rotation temperatures (T_{rot}), partition functions ($Q(T_{\text{rot}})$), column densities (N) and abundances (X) relative to H_2 . Abundances were calculated assuming a spherical envelope with the radius and number density $n(\text{H}_2)$ given in Gomez et al. (1991)

Species	T_{rot} (K)	$Q(T_{\text{rot}})$	N (cm^{-2})	$X(\text{H}_2)$
CH_3CN	65 ± 19	3151	$1.1 \pm 0.6 \cdot 10^{15}$	$3.6 \cdot 10^{-7}$
CH_3CCH	66 ± 6	575	$7.4 \pm 1.4 \cdot 10^{15}$	$2.4 \cdot 10^{-7}$
$\text{CH}_3\text{OH A}$	60 ± 4	590	$1.7 \pm 0.2 \cdot 10^{15}$	$5.6 \cdot 10^{-7}$
$\text{CH}_3\text{OH E}$	38 ± 11	294	$2.1 \pm 1.1 \cdot 10^{15}$	$6.8 \cdot 10^{-7}$
SO_2	111 ± 14	1378	$4.0 \pm 0.9 \cdot 10^{15}$	$1.3 \cdot 10^{-6}$
$^{34}\text{SO}_2$	154 ± 25	2212	$8.4 \pm 2.1 \cdot 10^{14}$	$2.8 \cdot 10^{-7}$
HC_3N	72 ± 15	996	$6.6 \pm 2.1 \cdot 10^{14}$	$2.2 \cdot 10^{-7}$

Table 4. Lower limits to column density. The partition function Q evaluated at T_{ex} is also given, where $T_{\text{ex}} = E_{\text{u}}/k$ for linear molecules and $\frac{2}{3}E_{\text{u}}/k$ for symmetric and asymmetric tops. Where two lines were used to evaluate the lower limit the partition functions of both lines are given

Species	$Q(T_{\text{ex}})$	N_{min} (cm^{-2})	Species	$Q(T_{\text{ex}})$	N_{min} (cm^{-2})
^{13}CO	12	$1.1 \cdot 10^{17}$	HCN	20	$6.0 \cdot 10^{14}$
CS	56	$4.6 \cdot 10^{14}$	HC^{15}N	20	$9.4 \cdot 10^{12}$
C^{34}S	56	$1.18 \cdot 10^{14}$	HN^{13}C	20	$8.0 \cdot 10^{12}$
HCO^+	20	$1.63 \cdot 10^{14}$	C_2H	40	$2.1 \cdot 10^{15}$
H^{13}CO^+	20	$1.94 \cdot 10^{13}$	OCS	812/870	$3.47 \cdot 10^{14}$
HC^{18}O^+	20	$2.1 \cdot 10^{12}$	SO	413/235	$4.1 \cdot 10^{15}$
H_2CS	1118/687	$1.5 \cdot 10^{15}$	^{34}SO	231/224	$3.0 \cdot 10^{14}$

peak $T_{\text{A}}^* \sim 0.5$ K (compared to the rms noise of this survey of 0.1 K at 1.25 MHz resolution). However there are several lines in the frequency range of the survey that exhibited strong emission from G34.26 (peak $T_{\text{A}}^* \geq 0.8$ K) and are noticeably absent in this survey. More sensitive observations targeted toward these species may be necessary to confirm that their absence is due to chemical processes.

The temperature of the gas associated with G5.89 is 60 – 80 K, as shown by the rotation diagrams of CH_3CN , CH_3CCH and HC_3N . This is reasonably consistent with the temperature of the molecular envelope (90 K) derived from NH_3 by Gomez et al. (1991). SO_2 and $^{34}\text{SO}_2$ appear to probe hotter gas with temperatures of 110 – 150 K. $^{34}\text{SO}_2$ has a higher rotation temperature which is consistent with a temperature gradient increasing toward the centre of the gas. The column densities of most species are in the range $10^{14} - 10^{15} \text{ cm}^{-2}$. The temperature and density of many of the species in this survey are much less than the corresponding species in the surveys of G34.26 (Macdonald et al. 1996) and Orion-KL (Schilke et al. 1997b). The most striking difference is for CH_3OH , which in G34.26 and Orion-KL has a rotation temperature of 340 and 190 K respectively but a rotation temperature of only 65 K in G5.89.

Inspection of the line profiles indicates that the chemistry of the molecular gas proceeds in two very different physical regimes: that in the molecular envelope surrounding the UC HII and that in the shocked gas of the molecular outflow. In the former the species tracing the envelope

are those such as CH_3CN , CH_3CCH and HC_3N ; in the latter sulphur- and silicon-bearing species with broad line profiles perhaps trace the chemistry of the outflow. Here, we will examine the broad details of the survey in order to set out the most useful approaches to investigate the chemistry of G5.89–0.39. The approaches outlined here will be elaborated upon more quantitatively in Paper II.

The chemistry of the envelope could be explained in terms of hot core chemical models (e.g. Millar et al. 1997 and references therein). Hot core models follow the evolution of the gas-phase chemistry of evaporated grain-mantle ices, where the evaporation is assumed to be caused by the “switch-on” of a nearby high-mass star. The hot core is usually extremely compact and dense with n_{H_2} around $10^6 - 10^7 \text{ cm}^{-3}$. Temperatures can range from 100 – 250 K and the chemistry is characterised by anomalously high abundances of saturated species (e.g. H_2S and NH_3) and large molecules such as $\text{C}_2\text{H}_5\text{OH}$ and CH_3OCH_3 . The nature of the envelope surrounding G5.89 has some parallels with hot core chemistry; there are high column densities of warm (~ 90 K) ammonia in close association with the young O-star (or stars) powering the UC HII region. There are however some differences between the molecular envelope and the standard picture of hot cores. G5.89 has the one of the lowest abundances of ammonia as seen in the UC HII region sample of Cesaroni et al. (1992), by roughly a factor of 10. The H_2 density lies between a few 10^4 and 10^6 cm^{-3} (Gomez et al. 1991 and Cesaroni et al. 1992). The molecular gas does not exhibit any detectable emission from large molecules, unlike the more line-rich

sources from the UC HII region survey of Hatchell et al. (1998a).

Hot core models may be able to explain these differences from the standard picture. We have examined the model of Millar et al. (1997) which is tailored for G34.26+0.15. The model takes account of the cloud structure as derived from the radiative transfer modelling of HCO⁺ (Heaton et al. 1993) and CO (Little et al. 1994). The cloud in which G34.26 is embedded has three hierarchical components; a cold rarefied halo approximately 3.5 pc in radius, a compact core of radius 0.1 pc and an inner dense ultracompact core of radius 0.01 pc. The compact core quite closely matches the conditions in the envelope of G5.89 (the model temperature and H₂ density are ~ 80 K and 10^6 cm⁻³ respectively). The compact core model shows that the abundances of the heavy organic molecules decrease as the core ages, falling off sharply beyond roughly 10^5 years. The compact core model is able to reproduce the abundances observed in the survey to within a factor of 10, with the exception of CH₃CCH which is under-produced in the model by at least three orders of magnitude. The observations are consistent with a chemical evolution time of $\geq 10^5$ years, although such a direct comparison of the model which is tailored for G34.26 and the observed abundances which are dependent on the geometry of G5.89 must be viewed with caution.

This must be reconciled with the apparent youth of the UC HII region. Recent high resolution astrometric observations have revealed that the HII region is expanding at a rate consistent with a dynamical age of ~ 600 years (Acord et al. 1998). However the onset of the UC HII region may have been delayed by high gas and dust densities in the early phases of star formation and the gas accreting onto the massive protostar may have undergone heating prior to stellar “switch-on”, perhaps from dynamical sources such as accretion shocks or outflow shocks. Another problem which addressed is that in hot core models the abundance of sulphur-bearing species such as H₂S and SO decreases after roughly 10^3 years for H₂S and 10^4 years for SO. The abundance of SO₂ stays enhanced until roughly 10^6 years. The hot core model is capable of reproducing the observed SO₂ abundances of $10^{-6} - 10^{-7}$ at a temperature of 100 K (Hatchell et al. 1998b), however the broad line wings of SO and SO₂ lines show that there must be a connection between the outflowing gas and the sulphur species.

Shock-driven chemical models predict the enhancement of sulphur-bearing species (e.g. Pineau-des-Fôrets et al. 1993; Draine et al. 1983), particularly H₂S and SO₂. A shock passing through molecular gas will heat and compress the gas, permitting reactions with activation energy barriers to proceed. The reactions $O + H_2 \rightarrow OH \rightarrow H_2O$ and $S + H_2 \rightarrow SH \rightarrow H_2S$ rapidly process the available oxygen and sulphur into H₂O and H₂S. Reactions with OH and O₂ convert the H₂S into SO and SO₂, which peak in abundance further behind the shock than H₂S. The model

of Pineau-des-Fôrets et al. (1993) predicts an SO₂ abundance of $\sim 10^{-6}$ which is consistent with that observed.

The shocks also strip the ice mantles from dust grains, by thermal processes, sputtering (Schilke et al. 1997a) and grain-grain collisions (Caselli et al. 1997). Depending on the shock speed one or more of these processes may dominate overall and at high enough shock speeds grains may be completely disrupted (e.g. J-shocks with speeds of > 40 km s⁻¹, Seab & Schull 1983). This disruption of grain cores or mantles injects depleted species into the gas phase, altering the chemical composition significantly. Chemical models involving grain disruption have been constructed for SiO and H₂O (e.g. Schilke et al. 1997a; Bergin et al. 1998) but not explicitly for sulphur-bearing species. Significant quantities of H₂S may be injected into the gas phase directly from grain mantles, rather than by hydrogenation of elemental sulphur.

It is also difficult to distinguish the route towards the production of sulphur-bearing species. Hot core chemistry alone can reproduce the observed abundances of sulphur-bearing species (Charnley 1997; Hatchell et al. 1998b). The chemical pathways are similar but for the formation of H₂S, which instead of being formed by hydrogenation of elemental sulphur is directly evaporated from grain ice mantles. However the survey shows that many sulphur-bearing species are likely to be located within the outflowing gas, by virtue of their broad line profiles. We intend to explore the mechanisms behind the sulphur species production further by detailed physical analysis and modelling of hot core and shock-driven chemistry in the companion paper (Paper II).

5. Summary and conclusions

We have performed a 330 – 360 GHz molecular line survey of the molecular gas associated with the UC HII region G5.89–0.39. The lines detected in the survey can be split into two types; those with narrow line profiles which probe the dense molecular envelope surrounding the UC HII region and those with broad line wings which probably arise from the massive outflow known to be associated with G5.89–0.39. The species which exhibit high-velocity emission are consistent with those seen toward outflows from low-mass YSOs, indicating that similar processes are responsible for the chemical evolution of outflowing gas. The rotation temperature of the species detected in the survey is 60 – 80 K and the column densities of most species are in the range $10^{14} - 10^{15}$ cm⁻².

No heavy organic species (e.g. CH₃OCH₃, CH₂CHCN) were detected. The spectrum is dominated by lines of sulphur-bearing species; over 50% of the identified lines originate from sulphur-bearing species. The dominant molecule in terms of integrated intensity is CO, which has twice the integrated intensity of SO₂. This is the opposite of the situation in Orion-KL (Schilke et al. 1997b) where SO₂ has twice the integrated intensity of CO. Beam dilution is the most likely cause.

The chemistry of G5.89 has features that are similar to hot core and shock-driven models. The broad features of the envelope chemistry are similar to those of hot cores with the caveats that the density and temperature are lower than in other hot core sources and that there is no evidence of the heavy organic molecules seen in the hot core surveys of Macdonald et al. (1996), Schilke et al. (1997b) and Hatchell et al. (1998a). It is possible that the envelope of G5.89 is evolved much further than the dynamical ages of both the UC HII region and outflow suggest. We intend to subject the survey data to detailed physical and chemical analysis in a following paper, in order to derive the physical parameters of the molecular gas and to model the chemistry of the envelope and outflow to determine whether the chemical evolution can be better described by hot core or shock-driven models.

Interferometric observations would also be extremely useful in determining the locations of the sulphur-bearing species and to confirm whether shock-driven chemistry or hot core chemistry is the prime cause. Acord et al. (1997) remark upon the similarity of the line profiles of SiO, NH₃ and CO at high velocities, perhaps suggesting that these three species are coexistent in the outflow. Interferometric observation of these species would confirm this and allow the refinement of models of the chemistry and physics of outflow shocks.

Acknowledgements. MAT would like to thank PPARC for their support via a research studentship and the JCMT staff for all their assistance during the observations.

References

- Acord J.M., Walmsley C.M., Churchwell E., 1997, ApJ 475, 693
 Acord J.M., Churchwell E., Wood D.O.S., 1998, ApJ 495, 107
 Anderson T., Herbst E., Delucia F.C., 1993, J. Mol. Spectrosc. 159, 410
 Bachiller R., 1996, ARA&A 34, 111
 Bachiller R., Perez-Gutierrez M., 1997, ApJ 487, L93
 Bergin E.A., Melnick G.J., Neufeld D.A., 1998, ApJ 499, 777
 Blake G.A., Sandell G., van Dishoeck E.F., Groesbeck T.D., Mundy L.G., Aspin C., 1995, ApJ 441, 689
 Cabrit S., Bertout C., 1992, A&A 261, 274
 Caselli P., Hartquist T.W., Havnes O., 1997, A&A 322, 296
 Cesaroni R., Walmsley C.M., Churchwell E., A&A 256, 618
 Charnley S.B., 1997, ApJ 481, 396
 Churchwell E., 1997, ApJ 479, L59
 De Young D.S., 1986, ApJ 263, L73
 Draine B.T., Roberge W.G., Dalgarno A., 1983, ApJ 264, 485
 Fukui Y., Iwata T., Mizuno A., Bally J., Lane A.P., 1993, in Protostars & Planets III, Levy E.H. & Lunine J.I. (eds.). Tucson: Univ. Ariz. Press, pp. 603–39
 Gomez Y., Rodriguez L.F., Garay G., Moran J.M., 1991, ApJ 377, 519
 Harvey P.M., Forveille T., 1988, A&A 197, L19
 Harvey P.M., Lester D.F., Colomé C., et al., 1994, ApJ 433, 187
 Hatchell J., Thompson M.A., Millar T.J., Macdonald G.H., 1998a, A&AS (in press)
 Hatchell J., Thompson M.A., Millar T.J., Macdonald G.H., 1998b, A&A 338, 713
 Heaton B.D., Little L.T., Yamashita T., et al., 1993, A&A 278, 238
 Jewell P.R., Hollis J.M., Lovas F.J., Snyder L.E., 1989, ApJS 70, 833
 Kutner M.L., Ulich B.L., 1981, ApJ 250, 341
 Little L.T., Gibb A.G., Heaton B.D., Ellison B.N., Claude S.M.X., 1994, MNRAS 271, 649
 Macdonald G.H., Gibb A.G., Habing R.J., Millar T.J., 1996, A&AS 119, 333 (Paper I)
 Millar T.J., Macdonald G.H., Gibb A.G., 1997, A&A 325, 1163
 Pelletier G., Pudritz R.E., 1992, ApJ 394, 117
 Pineau-des-Fôrets G., Roueff E., Schilke P., Flower D.R., 1993, MNRAS 262, 915
 Plume R., Jaffe D.T., Evans N.J. II, Martin-Pintado J., Gomez-Gonzalez J., 1997, ApJ 476, 730
 Poynter R.L., Pickett H.M., 1985, Appl. Opt. 24, 2335
 Schilke P., Walmsley C.M., Pineau-de-Fôrets G., Flower D.R., 1997a, A&A 321, 293
 Schilke P., Groesbeck T.D., Blake G.A., Phillips T.G., 1997b, ApJS 108, 301
 Seab C.G., Shull J.M., 1983, ApJ 275, 652
 Shepherd D.S., Churchwell E., 1996a, ApJ 457, 267
 Shepherd D.S., Churchwell E., 1996b, ApJ 472, 225
 Shepherd D.S., Watson A.M., Sargent A.I., Churchwell E., 1998, ApJ (in press)
 Shu F.H., Najita J., Ostriker E.C., Shang H., 1995, ApJ 455, L155
 Wood D.O.S., Churchwell E., 1989, ApJS 69, 831

Table 5. The measured line parameters of observed frequency ($\nu(\text{obs})$), peak temperature (T_A^*) and line width ($\Delta\nu_{1/2}$) for each detected line are listed here. Multiple detections of the same line have been included. Lines that are blended are indicated in the Notes column by *blended* if they are blended with a different species or *h/fines* if they are a mixture of two or more hyperfine components. Self-absorbed lines are also listed. Blended lines that can have one or both components extracted are indicated by *sl-blend*

$\nu(\text{obs})$ (GHz)	T_A^* (K)	$\Delta\nu_{1/2}$ (MHz)	Species	Transition	$\nu(\text{rest})$ (GHz)	E_u/k (K)	Notes
330.588	19.11	8.7	^{13}CO	3–2	330.588	31.7	
331.014	0.68	6.8	CH_3CN	18(3)–17(3)	331.0143	215	
331.045	0.70	5.8	CH_3CN	18(2)–17(2)	331.0461	180	
331.066	1.01		CH_3CN	18(1)–17(1)	331.0652	158	blend
331.072	0.87		CH_3CN	18(0)–17(0)	331.0716	151	blend
331.500	0.93	5.2	CH_3OH	11(1)–11(0) A–, A+	331.5024	169	
331.502	0.93	8.9	CH_3OH	11(1)–11(0) A–, A+	331.5024	169	
331.581	0.93	2.9	SO_2	11(6,6)–12(5,7)	331.5803	149	
331.582	0.73	7.3	SO_2	11(6,6)–12(5,7)	331.5803	149	
332.091	2.23	6.4	SO_2	21(2,20)–21(1,21)	332.0914	220	
332.092	1.76	6.4	SO_2	21(2,20)–21(1,21)	332.0914	220	
332.093	1.86	8.4	SO_2	21(2,20)–21(1,21)	332.0914	220	
332.093	1.09	7.4	SO_2	21(2,20)–21(1,21)	332.0914	220	
332.506	2.61	8.7	SO_2	4(3,1)–3(2,2)	332.5053	31.3	
332.506	2.43	8.8	SO_2	4(3,1)–3(2,2)	332.5053	31.3	
332.571	0.47	3.4	U				
333.901	1.87	6.2	^{34}SO	7(8)–6(7)	333.901	79.9	
333.901	1.96	6.2	^{34}SO	7(8)–6(7)	333.901	79.9	
334.673	2.10	7.7	SO_2	8(2,6)–7(1,7)	334.6733	43.1	
334.674	2.50	7.4	SO_2	8(2,6)–7(1,7)	334.6733	43.1	
335.574	0.63		U				
335.581	1.50	7.1	CH_3OH	7(1)–6(1) A+	335.582	79	
335.582	1.72	6.8	CH_3OH	7(1)–6(1) A+	335.582	79	
336.089	1.37	7.8	SO_2	23(3,21)–23(2,22)	336.0892	276	
336.090	1.42	7.8	SO_2	23(3,21)–23(2,22)	336.0892	276	
336.520	1.40	6.1	HC_3N	37–36	336.5201	307	
336.520	1.35	4.9	HC_3N	37–36	336.5201	307	
336.554	0.79	6.1	SO	10(11)–10(10)	336.5538	143	
336.554	0.94	6.1	SO	10(11)–10(10)	336.5538	143	
336.670	0.75	6.1	SO_2	16(7,9)–17(6,12)	336.6696	245	
336.670	0.77	2.2	SO_2	16(7,9)–17(6,12)	336.6696	245	
336.670	1.24	6.1	SO_2	16(7,9)–17(6,12)	336.6696	245	
336.671	0.66	6.1	SO_2	16(7,9)–17(6,12)	336.6696	245	
337.062	4.72	6.1	C^{17}O	3–2	337.0611	32.4	blend
337.063	6.40		C^{17}O	3–2	337.0611	32.4	blend
337.198	0.69	6.5	^{33}SO	8(7)–7(6)	337.1950		
337.198	0.66	7.4	^{33}SO	8(7)–7(6)	337.1950		
337.396	4.87	9.5	C^{34}S	7–6	337.3966	64.8	
337.396	4.03	6.5	C^{34}S	7–6	337.3966	64.8	
337.397	3.99	6.1	C^{34}S	7–6	337.3966	64.8	
337.397	4.67	6.5	C^{34}S	7–6	337.3966	64.8	
337.581	2.11	72	^{34}SO	8(8)–7(7)	337.5802	86.1	
337.581	1.99	7.2	^{34}SO	8(8)–7(7)	337.5802	86.1	sl-blend
338.082	2.61	5.2	H_2CS	10(1,10)–9(1,9)	338.081	102	sl-blend
338.084	2.48	5.1	H_2CS	10(1,10)–9(1,9)	338.081	102	
338.084	2.48	5.2	H_2CS	10(1,10)–9(1,9)	338.081	102	sl-blend
338.084	2.30	5.1	H_2CS	10(1,10)–9(1,9)	338.081	102	
338.125	2.79	6.5	CH_3OH	7(0)–6(0) E	338.1245	76.9	
338.125	1.74	4.9	CH_3OH	7(0)–6(0) E	338.1245	76.9	
338.306	1.50	8.4	SO_2	18(4,14)–18(3,15)	338.306	197	sl-blend
338.306	1.25	8.5	SO_2	18(4,14)–18(3,15)	338.306	197	sl-blend

Table 5. continued

$\nu(\text{obs})$ (GHz)	T_A^* (K)	$\Delta\nu_{1/2}$ (MHz)	Species	Transition	$\nu(\text{rest})$ (GHz)	E_u/k (K)	Notes
338.321	0.67	7.2	$^{34}\text{SO}_2$	13(2,12)–12(1,11)	338.3204	92.4	sl-blend
338.321	0.75	10	$^{34}\text{SO}_2$	13(2,12)–12(1,11)	338.3204	92.4	sl-blend
338.345	2.63	7.4	CH_3OH	7(–1)–6(–1) E	338.3446	69.4	
338.345	2.74	6.1	CH_3OH	7(–1)–6(–1) E	338.3446	69.4	
338.409	3.00	6.5	CH_3OH	7(0)–6(0) A+	338.4087	65	
338.409	2.79	6	CH_3OH	7(0)–6(0) A+	338.4087	65	
338.513	1.45		CH_3OH	7(2)–6(2) A–	338.5129	103	blend
338.513	1.28	6.4	CH_3OH	7(2)–6(2) A–	338.5129	103	blend
338.542	1.40		CH_3OH	7(3)–6(3) A+	338.5408	115	blend
338.543	1.27	7.5	CH_3OH	7(3)–6(3) A+	338.5408	115	blend
338.584	0.62	6.1	CH_3OH	7(3)–6(3) E	338.5832	112	
338.614	2.86	10	CH_3OH	7(1)–6(1) E	338.615	84.9	
338.614	3.12	10	CH_3OH	7(1)–6(1) E	338.615	84.9	
338.640	0.96	7.3	CH_3OH	7(2)–6(2) A+	338.6399	103	
338.640	1.22	7.4	CH_3OH	7(2)–6(2) A+	338.6399	103	
338.722	2.30	7.5	CH_3OH	7(2)–6(2) E	338.7216	86.1	blend
338.723	2.31	7.7	CH_3OH	7(2)–6(2) E	338.7216	86.1	blend
338.786	0.53	8.1	$^{34}\text{SO}_2$	14(4,10)–14(3,11)	338.7858	134	
338.787	0.48	9	$^{34}\text{SO}_2$	14(4,10)–14(3,11)	338.7858	134	
339.341	1.84	6.1	SO	3(3)–3(2)	339.3415	25.5	
339.342	1.59	6.7	SO	3(3)–3(2)	339.3415	25.5	
339.477	0.59	3.6	CN	3–2 2.5 2.5–2.5 2.5	339.4759	16.3	
339.478	0.37	4.9	CN	3–2 2.5 2.5–2.5 2.5	339.4759	16.3	
339.518	0.56	3.9	CN	3–2 2.5 3.5–2.5 3.5	339.5167	16.3	
339.518	0.72	5.2	CN	3–2 2.5 3.5–2.5 3.5	339.5167	16.3	
339.857	2.24	7.4	^{34}SO	9(8)–8(7)	339.8573	77.3	sl-blend
339.857	2.29	7.4	^{34}SO	9(8)–8(7)	339.8573	77.3	
340.008	1.19	6.1	CN	3–2 2.5 2.5–1.5 2.5	340.0082	16.3	sl-blend
340.009	1.09	5.8	CN	3–2 2.5 2.5–1.5 2.5	340.0082	16.3	sl-blend
340.021	1.33		CN	3–2 2.5 1.5–1.5 1.5	340.0196	16.3	blend
340.033	4.23	10	CN	3–2 2.5 3.5–1.5 2.5	340.0316	16.3	sl-blend
340.052	1.24		C^{33}S	7–6	340.0527	65.3	blend
340.247	5.20	8.7	CN	3–2 3.5 4.5–2.5 3.5	340.2479	16.3	sl-blend
340.248	5.65	7.1	CN	3–2 3.5 4.5–2.5 3.5	340.2479	16.3	blend
340.265	1.64		CN	3–2 3.5 3.5–2.5 3.5	340.265	16.3	blend
340.265	1.52	7.8	CN	3–2 3.5 3.5–2.5 3.5	340.265	16.3	blend
340.316	0.87		SO_2	28(2,26)–28(1,27)	340.3165	392	blend
340.317	0.80	6	SO_2	28(2,26)–28(1,27)	340.3165	392	sl-blend
340.317	0.69	7.4	SO_2	28(2,26)–28(1,27)	340.3165	392	
340.317	0.66	8.7	SO_2	28(2,26)–28(1,27)	340.3165	392	
340.414	4.78	10	SO	7(8)–6(7)	340.7143	81.2	
340.449	0.56	3.6	OCS	28–27	340.4493	237	
340.450	0.50	6	OCS	28–27	340.4493	237	
340.587	21.10	8.7	^{13}CO	3–2	330.588	31.7	
340.632	0.65	4.8	HC^{18}O^+	4–3	340.6330	40.9	
340.632	0.82	3.8	HC^{18}O^+	4–3	340.6330	40.9	
340.715	5.13	10	SO	7(8)–6(7)	340.7143	81.2	
340.836	0.44	8.7	^{33}SO	8(8)–7(7)	340.8390		

Table 5. continued

$\nu(\text{obs})$ (GHz)	T_A^* (K)	$\Delta\nu_{1/2}$ (MHz)	Species	Transition	$\nu(\text{rest})$ (GHz)	E_u/k (K)	Notes
341.350	1.01	5.2	HCS ⁺	8–7	341.3501	73.7	
341.351	1.20		HCS ⁺	8–7	341.3501	73.7	blend
341.416	1.97	6.4	CH ₃ OH	7(1)–6(1) A–	341.4156	80.1	
341.416	2.18	7.7	CH ₃ OH	7(1)–6(1) A–	341.4156	80.1	sl-blend
341.638	0.53	2.6	CH ₃ CCH	20(4)–19(4)	341.6371	287	
341.683	1.15	4.8	CH ₃ CCH	20(3)–19(3)	341.6826	237	
341.683	1.22	4.1	CH ₃ CCH	20(3)–19(3)	341.6826	237	
341.716	1.07	5.2	CH ₃ CCH	20(2)–19(2)	341.7151	201	
341.716	1.31	5.2	CH ₃ CCH	20(2)–19(2)	341.7151	201	
341.735	1.64		CH ₃ CCH	20(1)–19(1)	341.7346	179	blend
341.736	1.68	4.6	CH ₃ CCH	20(1)–19(1)	341.7346	179	sl-blend
341.742	1.71		CH ₃ CCH	20(0)–19(0)	341.7411	172	blend
341.742	1.73	5.5	CH ₃ CCH	20(0)–19(0)	341.7411	172	sl-blend
342.208	0.63	5.3	³⁴ SO ₂	5(3,3)–4(2,2)	342.2089	35.1	sl-blend
342.208	0.72	7.2	³⁴ SO ₂	5(3,3)–4(2,2)	342.2089	35.1	
342.233	0.42	5.3	³⁴ SO ₂	20(1,19)–19(2,18)	342.2317	198	
342.233	0.56	5.2	³⁴ SO ₂	20(1,19)–19(2,18)	342.2317	198	
342.332	0.67	7.6	³⁴ SO ₂	12(4,8)–12(3,9)	342.3321	110	sl-blend
342.333	0.51	7.5	³⁴ SO ₂	12(4,8)–12(3,9)	342.3321	110	
342.882	12.01	12	CS	7–6	342.8829	65.8	self-abs
342.883	10.29	11	CS	7–6	342.8829	65.8	self-abs
342.947	1.09	9.3	H ₂ CS	10(0,10)–9(0,9)	342.9443	90.6	
342.948	1.08	4.9	H ₂ CS	10(0,10)–9(0,9)	342.9443	90.6	
343.088	0.70	2.3	³³ SO	8(9)–7(8)	343.0860		
343.088	0.74	7.8	³³ SO	8(9)–7(8)	343.0860		
343.324	1.27	8.7	H ₂ CS	10(2,9)–9(2,8)	343.3198	143	*
			H ₂ ¹³ CO	5(1,5)–4(1,4)	343.3257	61.3	
343.324	1.24	6.8	H ₂ CS	10(2,9)–9(2,8)	343.3198	143	*
			H ₂ ¹³ CO	5(1,5)–4(1,4)	343.3257	61.3	
343.413	1.06	9.4	H ₂ CS	10(3,8)–9(3,7)	343.4077	209	blend
343.413	1.15	6.1	H ₂ CS	10(3,8)–9(3,7)	343.4077	209	blend
343.814	0.69	7.4	H ₂ CS	10(2,8)–9(2,7)	343.8109	143	
343.815	0.81	4.1	H ₂ CS	10(2,8)–9(2,7)	343.8109	143	
344.200	1.41	6.5	HC ¹⁵ N	4–3	344.2003	41.3	
344.200	1.81	5.4	HC ¹⁵ N	4–3	344.2003	41.3	
344.246	0.60	4.8	³⁴ SO ₂	10(4,6)–10(3,7)	344.2454	88.4	
344.310	6.05	8.8	SO	8(8)–7(7)	344.3107	87.5	
344.311	6.07	11	SO	8(8)–7(7)	344.3107	87.5	
344.969	0.69	7.7	SO ₂	40(4,36)–40(3,37) v ₂ = 1	344.9742	1550	*
			C ₂ H ₅ OH	21(0,21)–20(1,20)	244.9676	241	
344.986	0.55	9.7	³⁴ SO ₂	15(4,12)–15(3,13)	344.9876	148	sl-blend
344.990	0.94	8.7	³⁴ SO ₂	15(4,12)–15(3,13)	344.9876	148	sl-blend
344.997	0.60		³⁴ SO ₂	11(4,8)–11(3,9)	344.9982	98.5	blend
344.998	0.59	10	³⁴ SO ₂	11(4,8)–11(3,9)	344.9982	98.5	sl-blend
345.149	0.44	6.4	SO ₂	5(5,1)–6(4,2)	345.1491	75.1	
345.149	0.36	5.5	SO ₂	5(5,1)–6(4,2)	345.1491	75.1	
345.169	0.57	6.1	³⁴ SO ₂	8(4,4)–8(3,5)	345.1688	70.9	
345.169	0.73	7.4	³⁴ SO ₂	8(4,4)–8(3,5)	345.1688	70.9	
345.285	0.63	7.5	³⁴ SO ₂	9(4,6)–9(3,7)	345.2857	79.2	
345.287	0.53	6.2	³⁴ SO ₂	9(4,6)–9(3,7)	345.2857	79.2	
345.339	6.15	8.8	SO ₂	13(2,12)–12(1,11)	345.3385	93	*
			H ¹³ CN	4–3	345.3398	41.3	
345.339	5.90	8.8	SO ₂	13(2,12)–12(1,11)	345.3385	93	*
			H ¹³ CN	4–3	345.3398	41.3	
345.520	0.58	4.8	³⁴ SO ₂	7(4,4)–7(3,5)	345.5198	63.6	
345.553	0.522	4.2	³⁴ SO ₂	6(4,2)–6(3,3)	345.5532	57.2	

Table 5. continued

ν (obs) (GHz)	T_A^* (K)	$\Delta\nu_{1/2}$ (MHz)	Species	Transition	ν (rest) (GHz)	E_u/k (K)	Notes
345.553	0.55	7.1	$^{34}\text{SO}_2$	6(4,2)–6(3,3)	345.5532	57.2	
345.591	0.40	5.8	$^{34}\text{SO}_2$	7(4,4)–7(3,5)	345.5198	63.6	sl-blend
345.608	1.00	6.1	HC_3N	38–37	345.6090	323	
345.609	0.97	6.1	HC_3N	38–37	345.6090	323	
345.651	0.54	5.8	$^{34}\text{SO}_2$	5(4,2)–5(3,3)	345.6514	51.7	
345.651	0.50	2.6	$^{34}\text{SO}_2$	5(4,2)–5(3,3)	345.6514	51.7	
345.794	26.74	45	CO	3–2	345.796	33.2	self-abs
345.794	25.11	44	CO	3–2	345.796	33.2	self-abs
345.929	0.56	10	$^{34}\text{SO}_2$	17(4,14)–17(3,15)	345.9294	179	
346.528	7.06	13	SO	9(8)–8(7)	346.5286	78.8	
346.528	7.26	11	SO	9(8)–8(7)	346.5286	78.8	
346.652	2.68	10	SO_2	19(1,19)–18(0,18)	346.6522	168	
346.653	2.80	10	SO_2	19(1,19)–18(0,18)	346.6522	168	
346.999	4.54	6.1	H^{13}CO^+	4–3	346.9985	41.6	
346.999	5.48	6.1	H^{13}CO^+	4–3	346.9985	41.6	blend
347.323	1.23	10	SiO	8–7	347.3306	75.0	blend
347.332	1.56	10	SiO	8–7	347.3306	75.0	blend
347.740	0.52	6.2	U				$\text{U}_{347.7430}$ (Schilke)
348.116	0.80	10	$^{34}\text{SO}_2$	19(4,16)–19(3,17)	348.1176	213	
348.117	0.91	5.1	$^{34}\text{SO}_2$	19(4,16)–19(3,17)	348.1176	213	
348.341	1.18	6.3	HN^{13}C	4–3	348.3403	41.8	
348.341	1.56	6.3	HN^{13}C	4–3	348.3403	41.8	
348.388	1.08	8.7	SO_2	24(2,22)–23(3,21)	348.388	293	
348.390	1.31	8.2	SO_2	24(2,22)–23(3,21)	348.388	293	
348.534	1.87	5.3	H_2CS	10(1,9)–9(1,8)	348.5321	105	
348.535	2.11	4.3	H_2CS	10(1,9)–9(1,8)	348.5321	105	
349.108	0.49	6.5	CH_3OH	14(1)–14(0) A+	349.1070	260	
349.339	7.63		C_2H	4.5–3.5	349.3381	41.9	blend
349.339	5.73	6	C_2H	4.5–3.5	349.3381	41.9	sl-blend
349.392	1.18		CH_3CN	19(3)–18(3)	349.393	232	blend
349.393	1.39		CH_3CN	19(3)–18(3)	349.393	232	blend
349.400	4.76	6.5	C_2H	3.5–2.5	349.4006	41.9	sl-blend
349.400	4.98	5.2	C_2H	3.5–2.5	349.4006	41.9	sl-blend
349.401	4.23	6	C_2H	3.5–2.5	349.4006	41.9	sl-blend
349.401	4.38	6.5	C_2H	3.5–2.5	349.4006	41.9	sl-blend
349.427	1.07	3.2	CH_3CN	19(2)–18(2)	349.4266	196	
349.428	0.69	7.4	CH_3CN	19(2)–18(2)	349.4266	196	
349.447	1.12	5.5	CH_3CN	19(1)–18(1)	349.4467	175	sl-blend
349.449	1.17		CH_3CN	19(1)–18(1)	349.4467	175	blend
349.453	0.85	5.1	CH_3CN	19(0)–18(0)	349.4534	168	sl-blend
349.454	1.11		CH_3CN	19(0)–18(0)	349.4534	168	blend
349.604	0.40	5.8	U				
350.689	2.25	7.5	CH_3OH	4(0)–3(–1) E	350.6877	35.1	*
			NO	3.5 0.5–2.5 0.5	350.6895	36.1	h/fines
350.689	2.67	6.2	CH_3OH	4(0)–3(–1) E	350.6877	35.1	*
			NO	3.5 0.5–2.5 0.5	350.6895	36.1	h/fines
351.044	1.24	5.5	NO	3.5 4.5–2.5 3.5	351.0435	36.1	sl-blend
351.044	1.37	4.9	NO	3.5 4.5–2.5 3.5	351.0435	36.1	sl-blend
351.052	1.63	5.2	NO	3.5 3.5–2.5 2.5	351.0517	36.1	sl-blend
351.052	1.40	6.2	NO	3.5 3.5–2.5 2.5	351.0517	36.1	sl-blend
351.257	2.39	10	SO_2	5(3,3)–4(2,2)	351.2572	35.9	
351.257	2.66	9.1	SO_2	5(3,3)–4(2,2)	351.2572	35.9	

Table 5. continued

$\nu(\text{obs})$ (GHz)	T_A^* (K)	$\Delta\nu_{1/2}$ (MHz)	Species	Transition	$\nu(\text{rest})$ (GHz)	E_u/k (K)	Notes
351.633	0.55	7.5	HNCO	16(0,16)–15(0,15)	351.6335	143	h/fine blend
351.634	0.98	5.5	HNCO	16(0,16)–15(0,15)	351.6335	143	h/fine blend
351.769	9.47		H ₂ CO	5(1,5)–4(1,4)	351.7686	62.5	blend
351.769	9.16	8.7	H ₂ CO	5(1,5)–4(1,4)	351.7686	62.5	sl-blend
351.874	1.79	8.8	SO ₂	14(4,10)–14(3,11)	351.8739	136	
351.874	2.06	9.1	SO ₂	14(4,10)–14(3,11)	351.8739	136	
352.599	0.89	5.5	OCS	29–28	352.5996	254	
352.600	0.62	7.1	OCS	29–28	352.5996	254	
354.503	14.34	14	HCN	4–3	354.5055	42.5	sl-blend
354.503	15.44	15	HCN	4–3	354.5055	42.5	sl-blend
354.597	1.00	6.5	U				sba
354.697	1.02	7.8	HC ₃ N	39–38	354.6975	341	
354.697	0.89	3.8	HC ₃ N	39–38	354.6975	341	
354.697	0.81	7.5	HC ₃ N	39–38	354.6975	341	
354.698	0.90	5.5	HC ₃ N	39–38	354.6975	341	
355.046	1.72	8.5	SO ₂	12(4,8)–13(3,9)	355.0456	111	
355.046	1.87	10	SO ₂	12(4,8)–13(3,9)	355.0456	111	sl-blend
355.570	0.52	7.4	S ¹⁸ O	9(8)–8(7)	355.5736	93.1	
355.571	0.41	7.7	S ¹⁸ O	9(8)–8(7)	355.5736	93.1	
355.603	0.78	6	CH ₃ OH	13(0)–12(1) A+	355.6030	211	
355.604	0.81	6.1	CH ₃ OH	13(0)–12(1) A+	355.6030	211	
356.008	0.38	6.1	CH ₃ OH	15(1)–15(0) A–, A+	356.0066	296	
356.041	0.67	6.1	SO ₂	15(7,9)–16(6,10)	356.0407	230	
356.042	0.64	10	SO ₂	15(7,9)–16(6,10)	356.0407	230	
356.223	0.27	8.7	³⁴ SO ₂	25(3,23)–25(2,14)	356.2224	320	
356.733	22.41	12	HCO ⁺	4–3	356.7342	42.8	self-abs
356.733	21.61	13	HCO ⁺	4–3	356.7342	42.8	self-abs
357.165	1.73	9.1	SO ₂	13(4,10)–13(3,11)	357.1654	123	sl-blend
357.167	3.00		SO ₂	13(4,10)–13(3,11)	357.1654	123	blend
357.241	2.19	10	SO ₂	15(4,12)–15(3,13)	357.2412	150	sl-blend
357.387	1.62	10	SO ₂	11(4,8)–11(3,9)	357.3876	100	
357.387	1.72	10	SO ₂	11(4,8)–11(3,9)	357.3876	100	
357.581	1.45	10	SO ₂	8(4,4)–8(3,5)	357.5815	72.4	
357.582	1.56	8	SO ₂	8(4,4)–8(3,5)	357.5815	72.4	
357.660	0.74		U				sba blend
357.672	1.97	8.7	SO ₂	9(4,6)–9(3,7)	357.6719	80.6	
357.672	1.95	8.4	SO ₂	9(4,6)–9(3,7)	357.6719	80.6	
357.892	1.79	10	SO ₂	7(4,4)–7(3,5)	357.8925	65	
357.893	1.90	10	SO ₂	7(4,4)–7(3,5)	357.8925	65	
357.926	1.81	10	SO ₂	6(4,2)–6(3,3)	357.9259	58.6	
357.927	1.89	9.7	SO ₂	6(4,2)–6(3,3)	357.9259	58.6	
357.963	1.55	10	SO ₂	17(4,14)–17(3,15)	357.9629	180	
357.963	1.68	8.9	SO ₂	17(4,14)–17(3,15)	357.9629	180	
358.013	1.53	8.9	SO ₂	5(4,2)–5(3,3)	358.0132	53.1	
358.013	1.68	7.7	SO ₂	5(4,2)–5(3,3)	358.0132	53.1	
358.039	1.34	10	SO ₂	4(4,0)–4(3,1)	358.038	48.5	sl-blend
358.039	1.42	8.9	SO ₂	4(4,0)–4(3,1)	358.038	48.5	

Table 5. continued

$\nu(\text{obs})$ (GHz)	T_{A}^* (K)	$\Delta\nu_{1/2}$ (MHz)	Species	Transition	$\nu(\text{rest})$ (GHz)	E_{u}/k (K)	Notes
358.216	2.56	9.7	SO ₂	20(0,20)–19(1,19)	358.2157	185	
358.216	2.70	10	SO ₂	20(0,20)–19(1,19)	358.2157	185	
358.348	0.33	8.4	³⁴ SO ₂	23(4,20)–23(3,21)	358.3473	292	
358.605	1.57	8.9	CH ₃ OH	4(1)–3(0) E	358.6058	43.1	
358.606	1.74	6.8	CH ₃ OH	4(1)–3(0) E	358.6058	43.1	
358.646	0.49	5.8	CH ₃ CCH	21(5)–20(5)	358.6475	369	*
			S ¹⁸ O	9(8)–8(8)	358.6488	99.3	
358.647	0.41	8.9	CH ₃ CCH	21(5)–20(5)	358.6475	369	*
			S ¹⁸ O	9(8)–8(8)	358.6488	99.3	
358.710	0.25	5.5	CH ₃ CCH	21(4)–20(4)	358.7089	305	
358.756	0.66	7.8	CH ₃ CCH	21(3)–20(3)	358.7566	254	
358.757	0.89	6.1	CH ₃ CCH	21(3)–20(3)	358.7566	254	
358.791	0.69	6.2	CH ₃ CCH	21(2)–20(2)	358.7908	218	
358.791	0.64	7.8	CH ₃ CCH	21(2)–20(2)	358.7908	218	
358.812	0.76		CH ₃ CCH	21(1)–20(1)	358.8112	197	blend
358.812	0.99	4.9	CH ₃ CCH	21(1)–20(1)	358.8112	197	sl-blend
358.819	0.87	4.8	CH ₃ CCH	21(0)–20(0)	358.8181	189	sl-blend
358.819	0.58		CH ₃ CCH	21(0)–20(0)	358.8181	189	blend
358.988	0.66	6.8	³⁴ SO ₂	15(2,14)–14(1,13)	358.9880	119	
358.989	0.61	7.1	³⁴ SO ₂	15(2,14)–14(1,13)	358.9880	119	
359.771	1.58	7.7	SO ₂	19(4,16)–19(3,17)	359.7707	214	
359.772	1.10	11	SO ₂	19(4,16)–19(3,17)	359.7707	214	